

Pyrrolidine Based Analogs of 3'-Deoxythymidine

Karl-Heinz Altmann

Ciba-Geigy Ltd., Central Research Laboratories, R-1060.2.34, CH-4002 Basel

Abstract: *Pyrrolidine based analogs of 3'-deoxythymidine 1a,b and 2a,b have been prepared via N-acyliminium ion intermediates 3 and 4.*

Over the last few years there has been an ever increasing interest in the synthesis and the pharmacological properties of 2',3'-dideoxynucleosides,^{1,2} due to the ability of a number of these compounds to inhibit the growth of various types of viruses. Two of the most prominent examples are azidothymidine (AZT) and 2',3'-dideoxycytidine, which are both used in the treatment of AIDS.²

Among others, one of the approaches pursued to create more effective and especially more selective antiviral inhibitors has been the replacement of the tetrahydrofuran ring of 2',3'-dideoxynucleosides by a cyclopentane moiety, leading to carbocyclic nucleosides,³ or by other five-membered heterocyclic rings. The latter class includes compounds that are derived from tetrahydrothiophene,⁴ 1,3-dioxolane,^{5a} 1,3-oxathiolane,^{5b} isoxazole,⁶ and pyrrolidine.^{7,8} In all the *pyrrolidine* based analogs of 2',3'-dideoxynucleosides reported so far, the nitrogen replaces either C-1⁷ or C-3⁸ of a carbocyclic nucleoside; in contrast, no analogs are known in the literature where a nitrogen-atom is substituted for the oxygen-atom of the dideoxyribose moiety.⁹

This paper now describes the synthesis of pyrrolidine based analogs of α - and β -3'-deoxythymidine **1a,b** and **2a,b** as the first examples of a potential new class of antiviral agents.



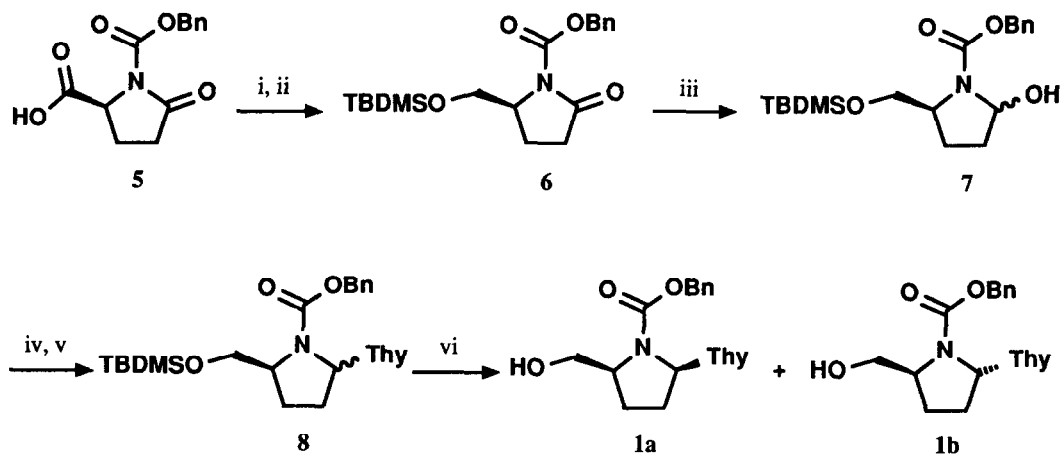
1: R = BnOCO (Z)] a: R¹ = thymine, R² = H
2: R = CH₃CO (Ac)] b: R¹ = H, R² = thymine

3: R = BnOCO
4: R = CH₃CO

The synthetic strategy applied is based on the Lewis acid catalyzed *in situ* generation of N-acyliminium ions¹⁰ **3** and **4** from the corresponding acetates and subsequent reaction of these key reactive intermediates with silylated thymine as a nucleophile.

The synthesis of Z-protected compounds **1a** and **1b** (Scheme 1) started from Z-protected (*S*)-pyroglutamic acid **5**¹¹ which was converted to the TBDMS protected alcohol **6**.¹² Selective reduction of

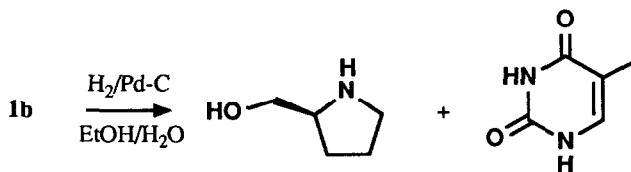
the lactam carbonyl group with $\text{NaBH}_4/\text{MeOH}$ ¹³ then gave the *N*-acylated acetal **7** in 95% yield as a mixture of diastereomers. Acetylation of **7** and subsequent reaction of the crude acetate with the bis-TMS derivative of thymine (generated *in situ* from thymine and *N,O*-bis-trimethylsilyl acetamide (BSA)) in the presence of SnCl_4 furnished protected nucleoside analog **8** in 64% yield as an inseparable mixture of diastereomers.^{14a} Desilylation of **8** with TBAF gave the target compounds **1a** and **1b** in 47% and 41% isolated yield, respectively.^{14b}



Scheme 1

i: a. $\text{EtOC(O)Cl}/N$ -methyl morpholine, THF, -5° , 5 min; b. NaBH_4 , 68%; ii: TBDMS-Cl, imidazole, DMF, quant.; iii: NaBH_4 (10 equiv.), MeOH, -5° , 30 min, 95%; iv. Ac_2O , DMAP (0.1 equiv.), $\text{CH}_2\text{Cl}_2/\text{pyridine}$; v. $(\text{TMS})_2$ -thymine, SnCl_4 , 3 h, -15° , 64% (2 steps); vi. TBAF, THF, **1a**: 47%, **1b**: 41%.

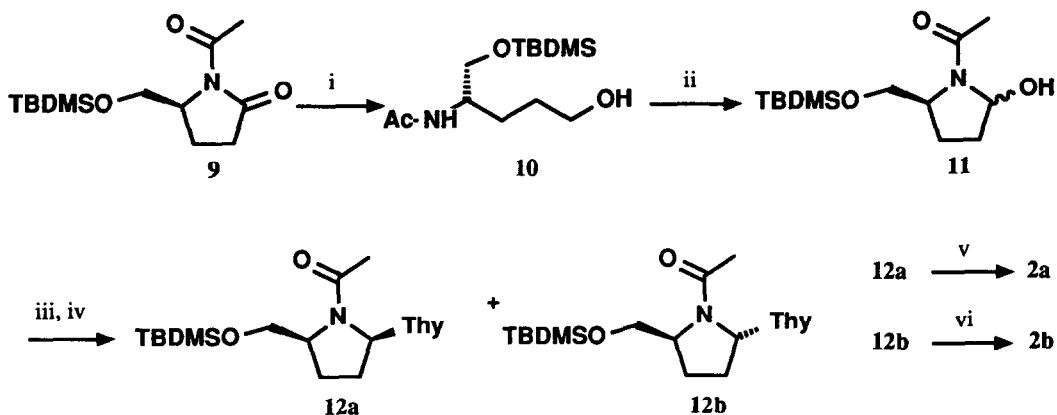
Catalytic hydrogenation of **1b** produced none of the deprotected dideoxynucleoside analog (with a free NH-group), but resulted in the formation of thymine and (*S*)-prolinol (Scheme 2).



Scheme 2

Presumably removal of the *Z*-protecting group from the pyrrolidine nitrogen is followed by rapid elimination of the base and the resulting imine is further hydrogenated to give (*S*)-prolinol.

The synthesis of dideoxynucleoside analogs **2a** and **2b** from acetylated lactam **9** (obtained from (*S*)-5-hydroxymethyl-2-pyrrolidinone (FLUKA) via silylation with TBDMS-Cl¹⁵ and subsequent acetylation with $\text{Ac}_2\text{O}/\text{DMAP}$; 80%) proved to be less straightforward than the preparation of **1a** and **1b** from **6** (Scheme 3).



Scheme 3

i: NaBH₄, MeOH, -5 °, 20 min, 48 %; ii: [n-C₃H₇]₄N(RuO₄) (0.1 equiv.), N-methyl morpholine N-oxide x H₂O (1.1 equiv.), mol. sieves, CH₂Cl₂, r.t., 3 h, 46 %; iii: Ac₂O, DMAP (0.1 equiv.), CH₂Cl₂/pyridine; iv: (TMS)₂-thymine, SnCl₄, CH₃CN, -15 °, 3 h, 55 % (**12a**: 21 %, **12b**: 31 %); v; vi: TBAF, THF, 64 %; 79 %.

In contrast to the N-Z-protected lactam **6** treatment of **9** with NaBH₄ under conditions that led to virtually quantitative conversion of **6** to **7** gave alcohol **10** as the major product in 48 % yield.¹⁶ Oxidation of **10** with tetra-n-propylammonium perruthenate (TPAP)/N-methyl morpholine N-oxide¹⁷ afforded the desired acetal **11** in 46 % yield, the moderate yield being due to rapid further oxidation of **11** to lactam **9**. If the oxidation is carried out with PCC¹⁸ or PDC¹⁹ the ratio **8/11** even increases and the reactions are generally less clean. Activation of **11** by acetylation and subsequent reaction of the crude acetate with (isolated) bis-TMS-thymine under SnCl₄ catalysis gave a 55 % yield of a *ca.* 2/3 mixture of diastereomers **12a** and **12b**, which could be separated by silica gel chromatography in AcOEt/MeOH to furnish **12a** and **12b** in 21 % and 31 % isolated yield, respectively. Interestingly, the stereochemical outcome of the reaction of bis-silylated thymine with N-acyliminium ion **4** does not conform to the pronounced "cis-effect" that is usually observed for related systems.²⁰

Desilylation of **12a** and **12b** produced the target compounds **2a** and **2b** in 64% and 78% yield, respectively.²¹

In summary, the synthesis of four representative examples of a new class of dideoxynucleoside analogs has been accomplished. The synthetic strategy outlined in this report for the synthesis of **1a,b** and **2a,b** is equally applicable to compounds of type **1** and **2** incorporating nucleic acid bases other than thymine. Moreover, this strategy has also been applied to the synthesis of the corresponding N-acetylated *thymidine* analog which may be an interesting building block for antisense oligonucleotides.²²

Acknowledgements

I thank Ms. G. Baisch for technical assistance and Dr. T. Winkler for the determination of α/β -ratios by ¹H-NMR spectroscopy.

References and Notes

1. a. Huryn, D.M.; Okabe, M. *Chem. Rev.* **1992**, *92*, 1745-1768. b. Dueholm, K.L.; Pedersen, E.B. *Synthesis* **1992**, 1-22.
2. Mitsuya, H.; Yarchoan, R.; Broder, S. *Science* **1990**, *249*, 1533-1544.
3. Borthwick, A.D.; Biggadike, K. *Tetrahedron* **1992**, *48*, 571-623.
4. Secrist, J.A., III; Riggs, R.M.; Tiwari, K.N.; Montgomery, J.A. *J. Med. Chem.* **1992**, *35*, 533-538.
5. a. Kim, H.O.; Schinazi, R.F.; Shanmuganathan, K.; Jeong, L.S.; Beach, J.W.; Nampalli, S.; Cannon, D.L.; Chu, C.K. *J. Med. Chem.* **1993**, *36*, 519-528. b. Jeong, L.S.; Schinazi, R.F.; Beach, J.W.; Kim, H.O.; Nampalli, S.; Shanmuganathan, K.; Alves, A.J.; McMillan, A.; Chu, C.K.; Mathis, R. *J. Med. Chem.* **1993**, *36*, 181-195.
6. Tronchet, J.M.J.; Iznaden, M.; Barbalat-Rey, F.; Dhimane, H.; Ricca, A.; Balzarini, J.; DeClercq, E. *Eur. J. Med. Chem.* **1992**, *27*, 555-560.
7. Harnden, M.R.; Jarvest, R.L. *J. Chem. Soc. Perkin Trans I* **1991**, 2073-2079.
8. a. Ng, K.E.; Orgel, L.E. *J. Med. Chem.* **1989**, *32*, 1754-1757. b. Peterson, M.L.; Vince, R. *J. Med. Chem.* **1991**, *34*, 2787-2797.
9. The synthesis of 1-(N-acetyl pyrrolidin-2-yl)-5-fluorouracil has been reported: Nishitani, T.; Horikawa, H.; Iwasaki, T.; Matsumoto, K.; Inuoue, I.; Miyoshi, M. *J. Org. Chem.* **1982**, *47*, 1706-1712.
10. For a review see: Hiemstra, H.; Speckamp, W.N. in "Comprehensive Organic Synthesis", Trost, B.M.; Fleming, I.; Eds.; Pergamon, Oxford 1991; Vol.2, Chapter 4.5.
11. Gibian, H.; Klieger, E. *Liebigs Ann. Chem.* **1961**, *640*, 145-156.
12. N-Z-(S)-5-hydroxymethyl-2-pyrrolidinone derived from the reduction of N-Z-(S)-pyroglutamic acid was also converted to the free pyrrolidinone by catalytic hydrogenation. The $[\alpha]_D^{25}$ of this material was + 23.5 °; comparison with the $[\alpha]_D^{25}$ of commercially available material (FLUKA) (+ 34.2 °) indicates an ee of 69 % which should be identical with the ee of **6**.
13. Miller, S.A.; Chamberlin, A.R. *J. Am. Chem. Soc.* **1990**, *112*, 8100-8112.
14. a. Due to the presence of rotamers, the $^1\text{H-NMR}$ spectrum of **8** at r.t. was completely uninformative with respect to the ratio of diastereomers. b. **1a**: $^1\text{H-NMR}$ (400 MHz, CD_3OD) δ 8.10 (s, br, 1H, H-6), 7.30 (m, br, 5H, H_{ar}), 6.15 (m, br, 1H, H-1'), 5.25 (m, br, 1H, $\text{CH}_2(\text{Z})$), 4.90 (m, br, 1H, $\text{CH}_2(\text{Z})$), 4.15 (m, br, 1H, H-5'), 3.95 (m, 1H, H-4'), 3.68 (d, J=3.0 Hz, 0.5H, H-5'), 3.66 (d, J=3.0 Hz, 0.5H, H-5'), 2.30 (m, 1H, H-2'), 2.10 (m, br, 3H, H-2' + H-3'), 1.80 (s, br, 3H, $\text{CH}_3(\text{Thy})$). **1b**: $^1\text{H-NMR}$ (400 MHz, CD_3OD) δ 7.40 - 7.15 (m, 5.33H, H_{ar} + H-6 (rotamer 1)), 7.00 (s, br, 0.67H, H-6 (rotamer 2)), 6.15 (d, J=8.0 Hz, 0.67 H, H-1'), 6.10 (d, J=6.5 Hz, 0.33H, H-1'), 5.35 (d, J=12.5 Hz, 0.67H, $\text{CH}_2(\text{Z})$), 5.15 (s, br, 0.67H, $\text{CH}_2(\text{Z})$), 4.80 (d, J=12.5 Hz, 0.67H, $\text{CH}_2(\text{Z})$), 4.25 (m, 1H, H-4'), 3.82 (d, J=3.0 Hz, 0.33H, H-5'), 3.78 (d, J=3.0 Hz, 0.33H, H-5'), 3.65 (m, 1.34H, H-5'), 3.65 (m, 1H), 2.30 - 2.15 (m, 1H), 2.10 - 2.00 (m, 1H), 1.95 - 1.85 (m, 1H), 1.85 (s, br, 1H, $\text{CH}_3(\text{Thy})$), 1.70 (s, br, 2H, $\text{CH}_3(\text{Thy})$).
15. Ackermann, J.; Matthes, M.; Tamm, C. *Helv. Chim. Acta* **1990**, *73*, 122-132.
16. Although treatment of **9** with NaBH_4/H^+ in EtOH (Hubert, J.C.; Wijnberg, J.B.P.A.; Speckamp, W.N. *Tetrahedron* **1975**, *31*, 1437-1441) in one experiment gave a 48 % yield of **11** (after chromatography) this result was not reproducible. In subsequent experiments the yield of **11** was much lower and the material was more difficult to purify than alcohol **10**.
17. Griffith, W.P.; Ley, S.V.; Whitcombe, G.P.; White, A.D. *J. Chem. Soc., Chem Comm.* **1987**, 1625-1627.
18. Corey, E.J.; Suggs, J.W. *Tetrahedron Lett.* **1975**, 2647-2650.
19. Corey, E.J.; Schmidt, G. *Tetrahedron Lett.* **1979**, 399-402.
20. **2a**: $^1\text{H-NMR}$ (250 MHz, $(\text{CD}_3)_2\text{SO}$) δ 8.19 (s, br, 0.55H, H-6 (rotamer 1)), 7.55 (s, br, 0.45H (rotamer 2)), 6.20 (m, br, 0.55H, H-1'), 5.95 (m, br, 0.45H, H-1'), 4.10 - 3.50 (m, br, 3H, H-5' + H-4'), 2.25 (m, br, 1H), 2.10 (s, br, 1.5H, $\text{CH}_3(\text{Ac})$), 2.05 - 1.80 (m, 4.5H, incl. s 1.85 ($\text{CH}_3(\text{Ac})$)), 1.75 (s, br, 3H, $\text{CH}_3(\text{Thy})$). **2b**: (400 MHz, CD_2OD) δ 7.27 (d, J<1 Hz, 0.46H, H-6 (rotamer 1)), 7.17 (d, J<1 Hz, 0.54H, H-6 (rotamer 2)), 6.29 (d, J=7.5 Hz, 0.54H, H-1'), 6.17 (d, J=7.5 Hz, 0.46H, H-1'), 4.45 - 4.35 (m, 1H, H-4'), 3.79 (dd, J=10.0 Hz, J=5.0 Hz, 0.54H, H-5'), 3.68 (dd, J=10.0 Hz, J=3.0 Hz, 0.54H, H-5'), 3.63 (m, 0.92H, H-5'), 2.66 - 2.48 (m, 1H), 2.39 - 2.28 (m, 0.46H), 2.38 - 2.12 (m, 2.9H, incl. s 2.22 ($\text{CH}_3(\text{Ac})$)), 2.07 - 1.85 (m, 5.7H, incl. s 1.98 ($\text{CH}_3(\text{Ac})$), s 1.90 ($\text{CH}_3(\text{Thy})$), s 1.85 ($\text{CH}_3(\text{Thy})$)).
21. a. Shono, T.; Matsumura, Y.; Tsubata, K.; Uchida, K. *J. Org. Chem.* **1986**, *51*, 2590-2592. b. Renaud, Ph.; Seebach, D. *Angew. Chem. Int. Ed. Engl.* **1986**, *25*, 843-844.
22. Altmann, K.-H.; Freier, S.; Pieleus, U.; Winkler, T., manuscript in preparation.

(Received in Germany 10 August 1993; accepted 29 September 1993)